

Supporting Information

Photoelectrochemical Oxidation of Water Using Nanostructured BiVO₄ Films

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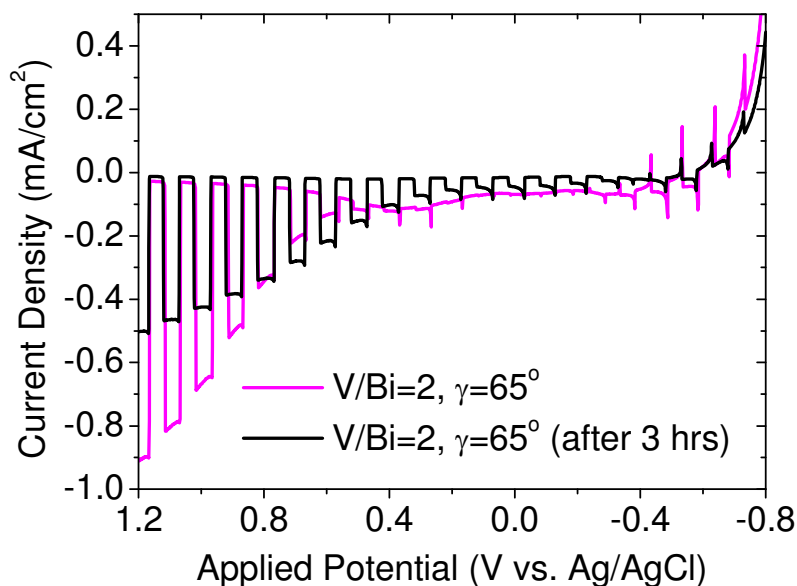


Figure S1. Chopped (dark and white light) LSV scans for a BiVO₄ film synthesized by RBD (V/Bi=2, $\gamma=65^\circ$, thickness=0.5 μm) in 0.5 M Na₂SO₄. The film was measured initially (magenta) and after illumination for 3 hours at a constant potential of 1.0 V vs. Ag/AgCl (black). The scan rate was 0.025 V/s.

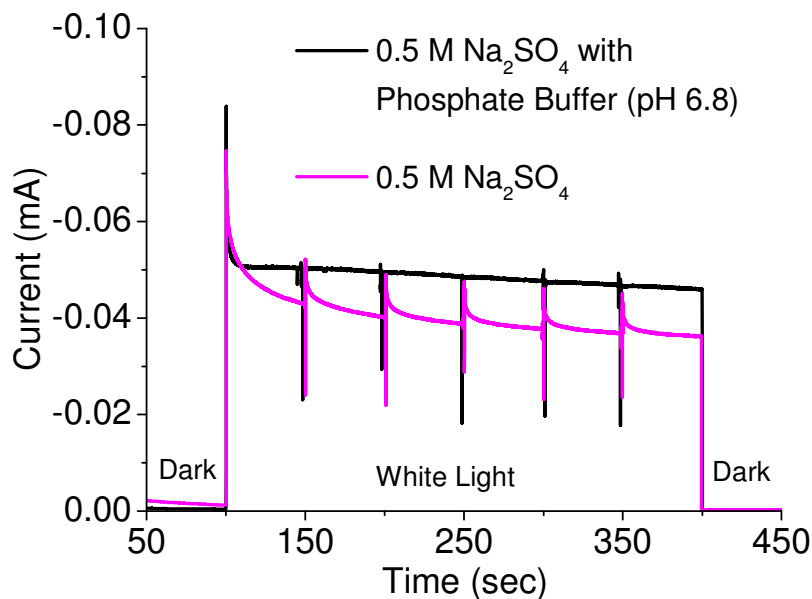


Figure S2. Amperometric *i-t* curve for a BiVO_4 film synthesized by RBD ($V/\text{Bi}=2$, $\gamma=65^\circ$, thickness= $0.5 \mu\text{m}$) in $0.5 \text{ M Na}_2\text{SO}_4$ (magenta) and $0.5 \text{ M Na}_2\text{SO}_4$ with 0.5 M phosphate buffer solution (pH 6.8) (black) at a constant applied potential of 1.0 V vs. Ag/AgCl . During white light illumination the film was flushed with fresh electrolyte every 50 seconds causing the spikes in photocurrent. For non-buffered $0.5 \text{ M Na}_2\text{SO}_4$ local pH changes near the film surface caused by the water oxidation reaction resulted in gradual photocurrent decay after each flush.

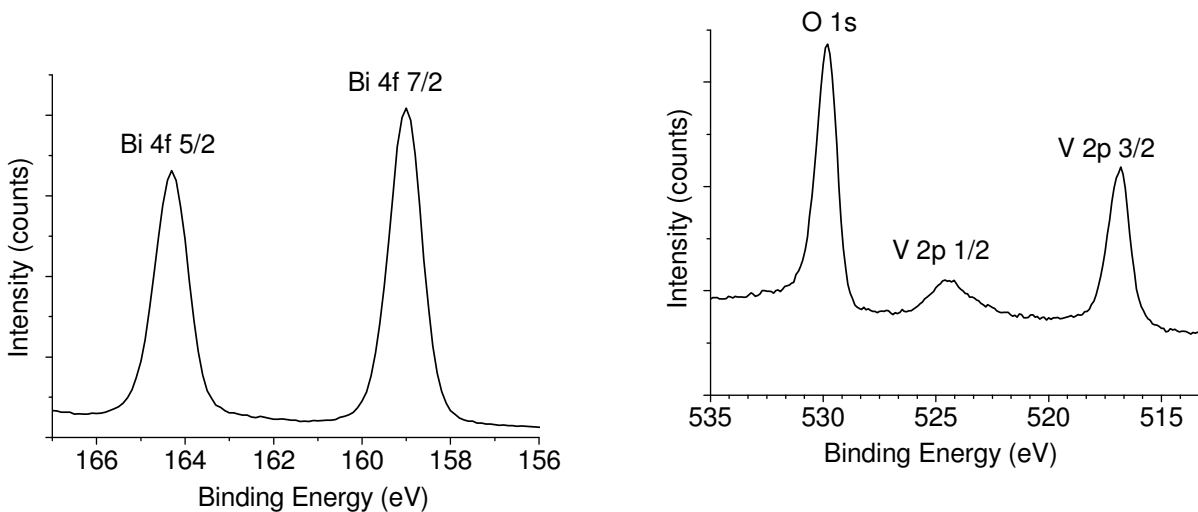


Figure S3. XPS spectra for BiVO_4 film synthesized by RBD ($V/\text{Bi}=2$, $\gamma=65^\circ$, thickness= $0.5 \mu\text{m}$) obtained before PEC testing. Binding energies of 159.0 eV for $\text{Bi } 4f \ 7/2$, 516.8 eV for $\text{V } 2p \ 3/2$, and 529.8 eV for $\text{O } 1s$ imply oxidation states of Bi^{3+} , V^{5+} , and O^{2-} .

Film Deposition Parameters and Surface Treatments	Ion Concentrations (ppb) in Unused Electrolyte		Ion Concentrations (ppb) in Used Electrolyte (3 hrs of illumination at 1.0V vs. Ag/AgCl)	
	Bi	V	Bi	V
V/Bi=2, $\gamma=65^\circ$	0.1	1.2	0.3	247
V/Bi=1, $\gamma=65^\circ$ (Co surface)	0.4	0.9	0.4	0.9

Table S1. ICP-MS results for electrolyte solutions used with two different BiVO₄ films synthesized by RBD. A batch of 0.1 M Na₂SO₄ was prepared for each film and measured to obtain the ion concentrations in unused electrolyte. About 25 mL of each 0.1 M Na₂SO₄ solution was used for PEC testing (over 3 hours of illumination at constant applied bias of 1.0 V vs. Ag/AgCl) and measured to obtain the ion concentrations in used electrolyte. All values were below the detection limits of the ICP-MS system except for the V concentration in used electrolyte from the V/Bi=2 film.